

High temperature morphology of phenolic resin pyrolysis

Collin Foster, Lincoln N. Collins, Scott A. Roberts, and Francesco Panerai

12th Ablation Workshop: November 9th, 2022



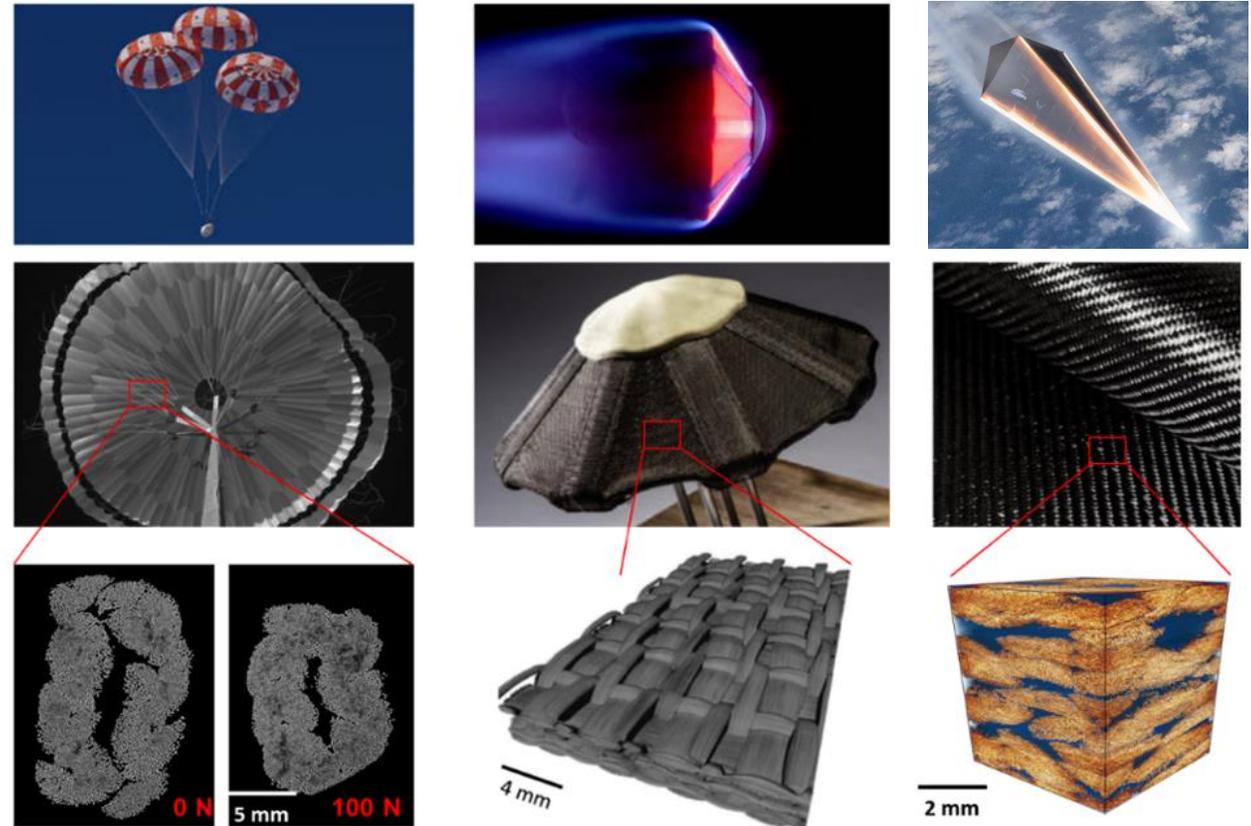
U.S. DEPARTMENT OF
ENERGY



Overview



1. X-ray computed tomography (XCT) at the Advanced Light Source (ALS)
2. Resin introduction, SC-1008
3. In-situ XCT results
4. Outlook

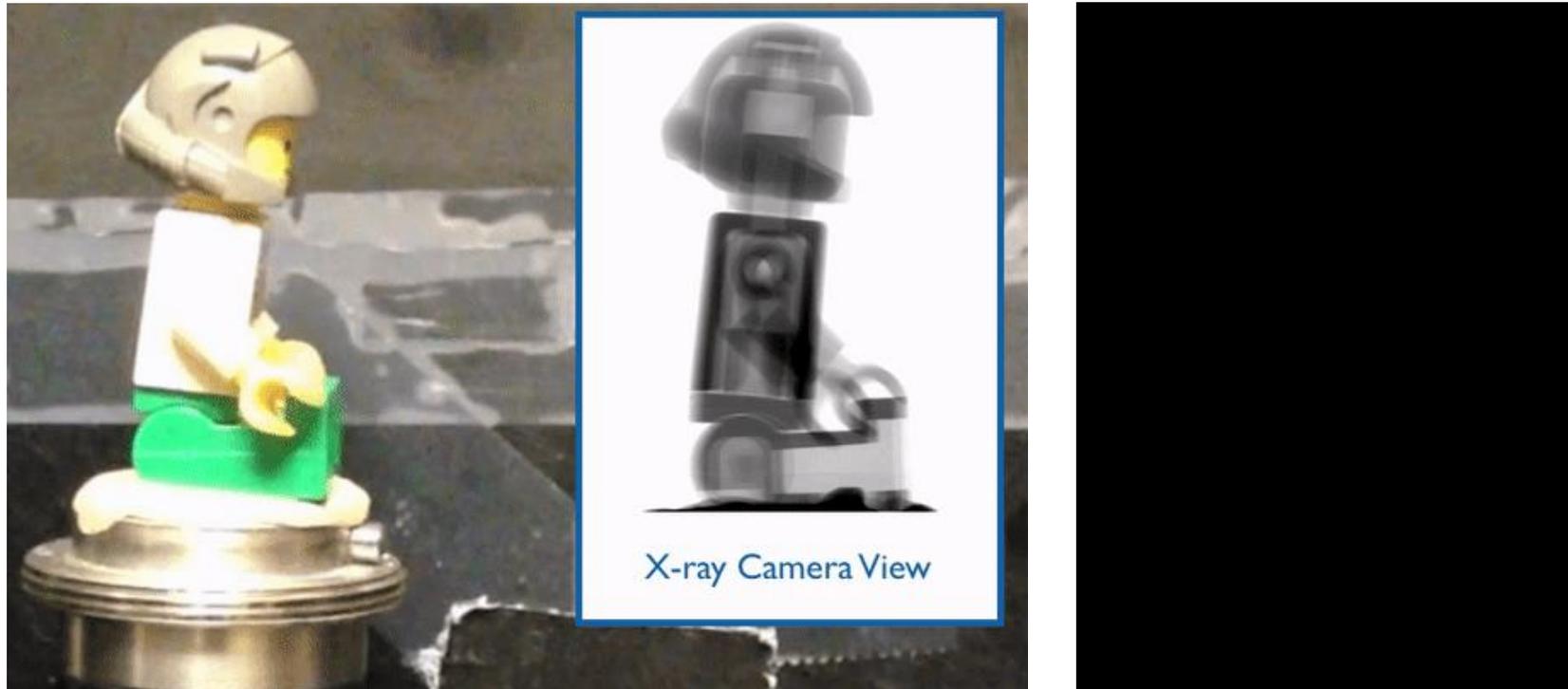


Aerospace material systems and XCT

XCT at the ALS



X-ray system	Resolution [$\mu\text{m}/\text{px}$]	Field of View, \varnothing [mm]	Time/tomography	Capabilities
Xradia Bio-MCT	0.9 – 10	1.5 – 10	> 2 hours	<i>ex-situ</i> , static scanning
ALS 8.3.2 Beamline	0.6 – 10	1.7 – 27	30 s – 30 min	<i>in-situ</i> , combined loading



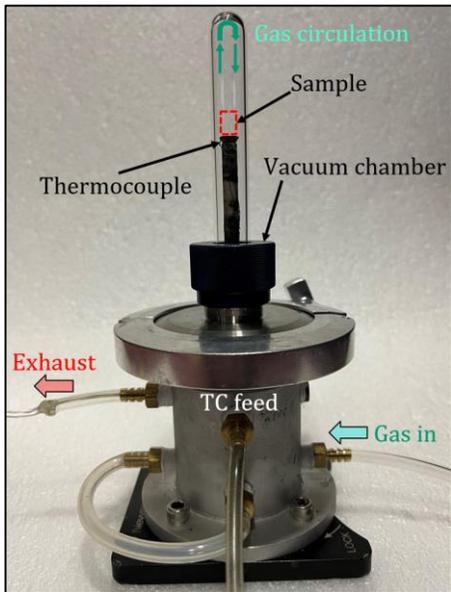
Synchrotron XCT of Astronaut

Courtesy of Dula Parkinson (Advanced Light Source)

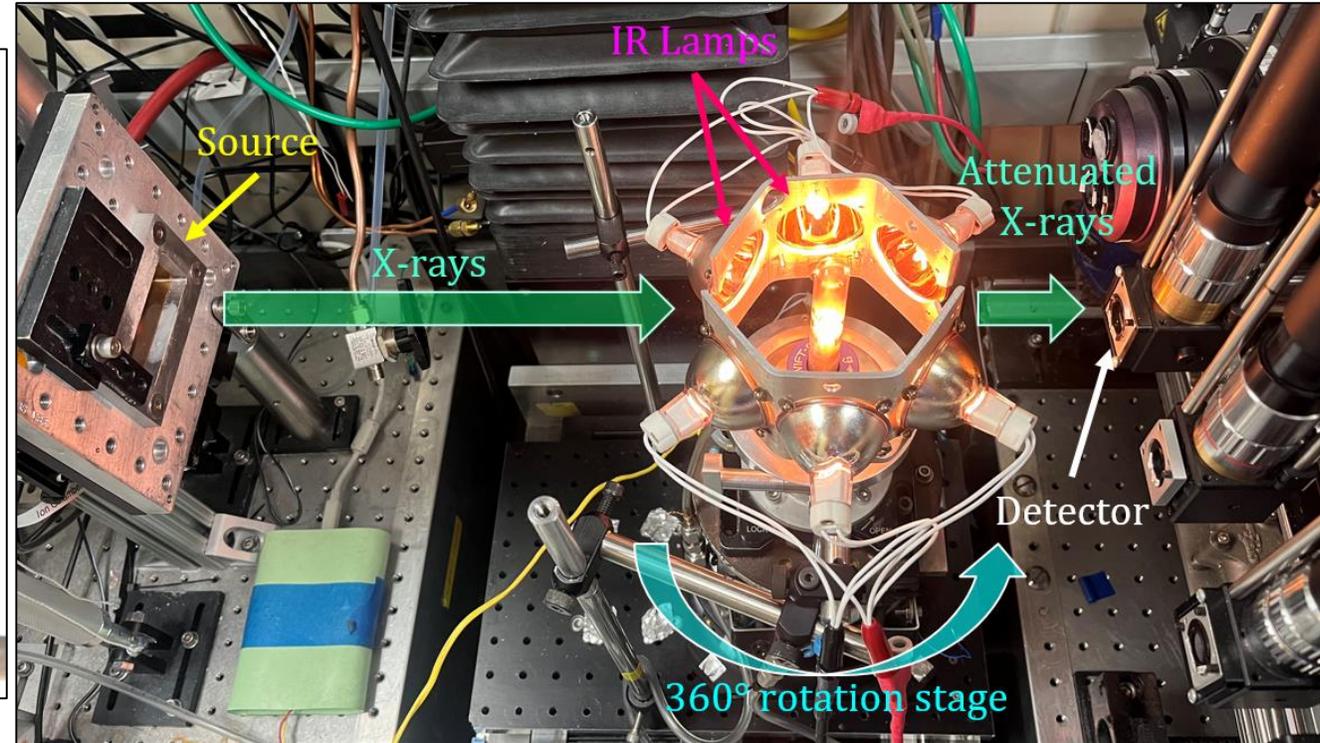
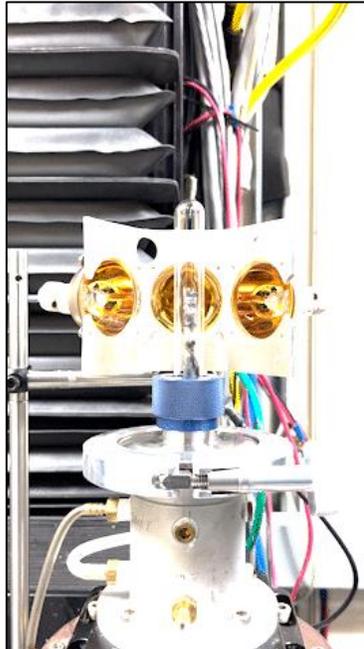
In-situ XCT at the ALS



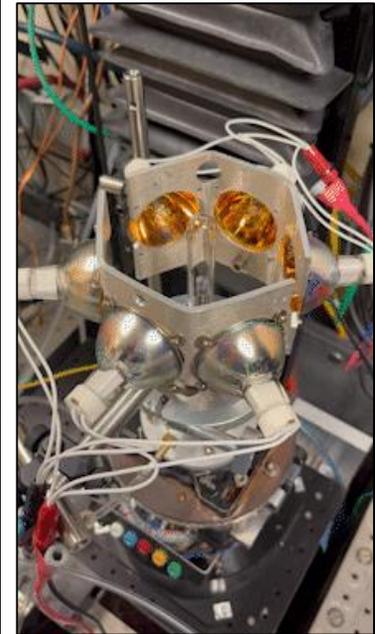
- ALS Synchrotron provides high flux of x-rays, allowing for *in-situ* XCT



Controlled-environment chamber



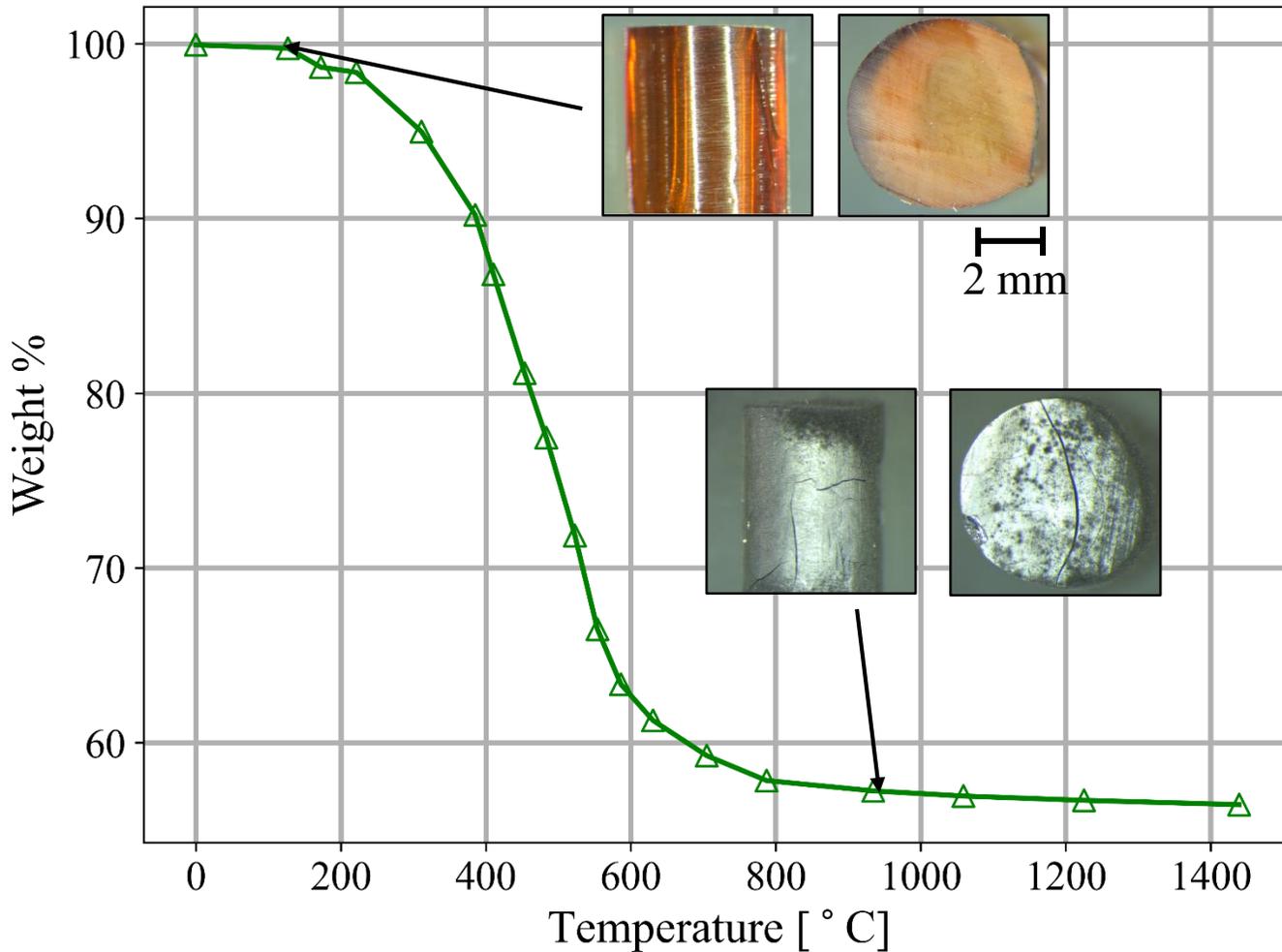
Inside XCT hutch at ALS



Resin systems



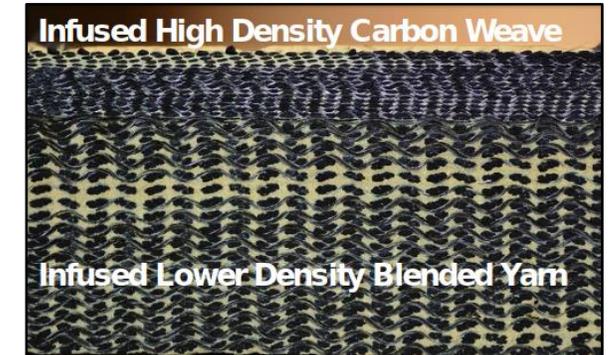
TGA SC-1008 phenolic resin



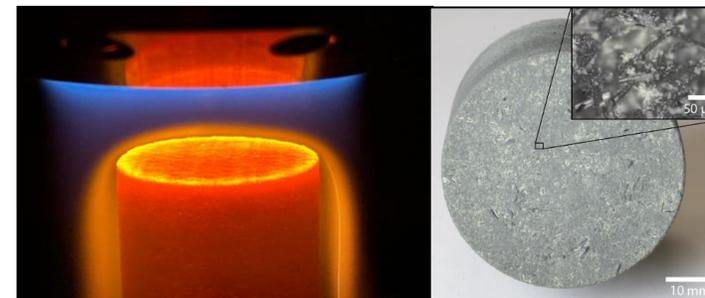
PICA
(Stackpole et al)



HEEET
(Ellerby et al)



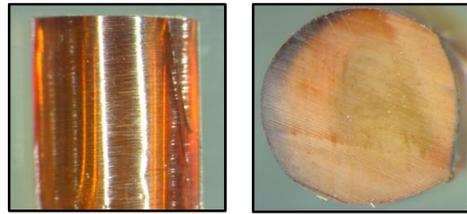
ADEPT
(Venkatapathy et al)



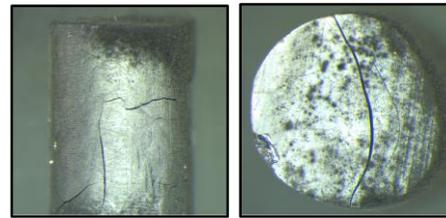
HARLEM
(Varona & Univ. of Stuttgart, HEFDiG)

[TGA: Hernandez-Sanchez, 2021]

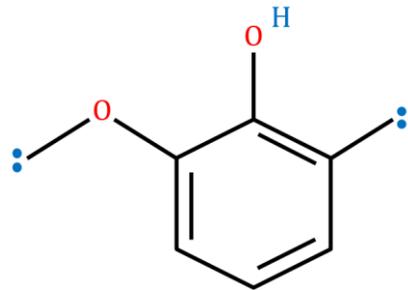
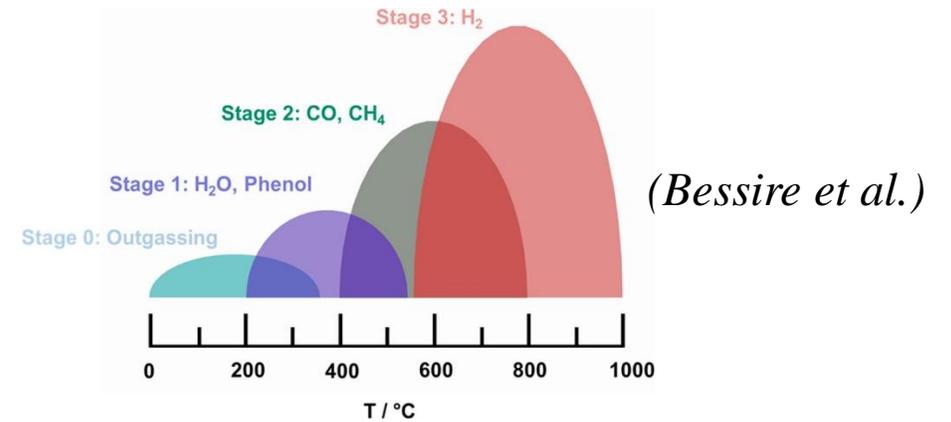
Resin chemistry



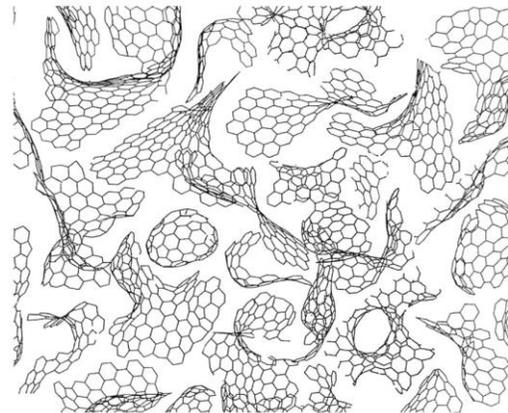
Virgin SC-1008



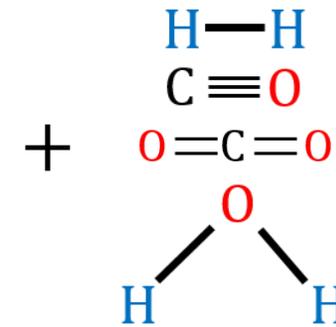
Pyrolyzed 1000 °C



Phenolic formaldehyde (s)

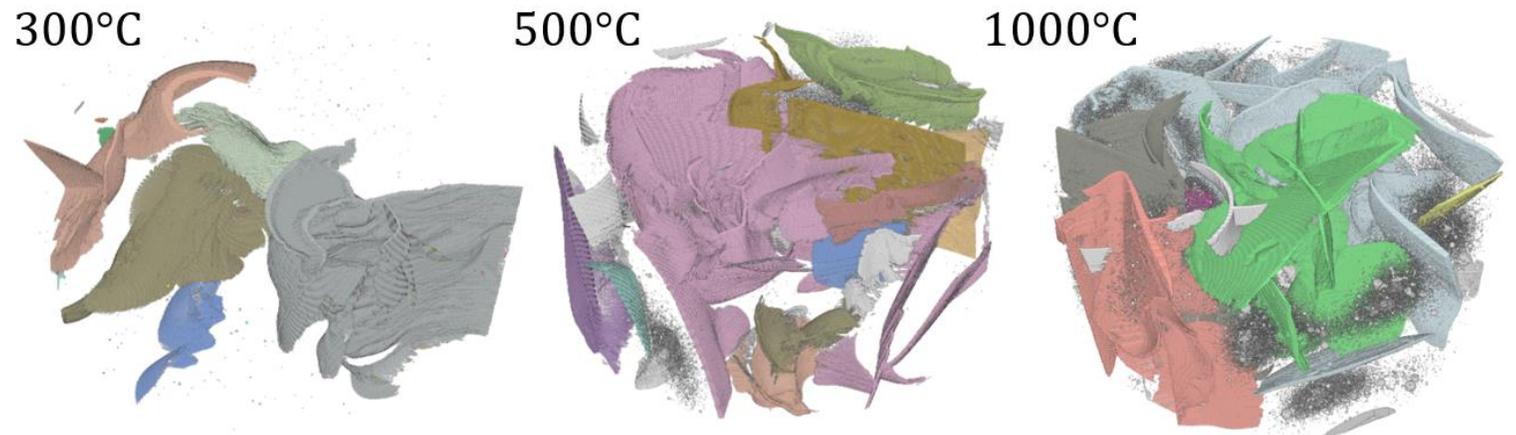
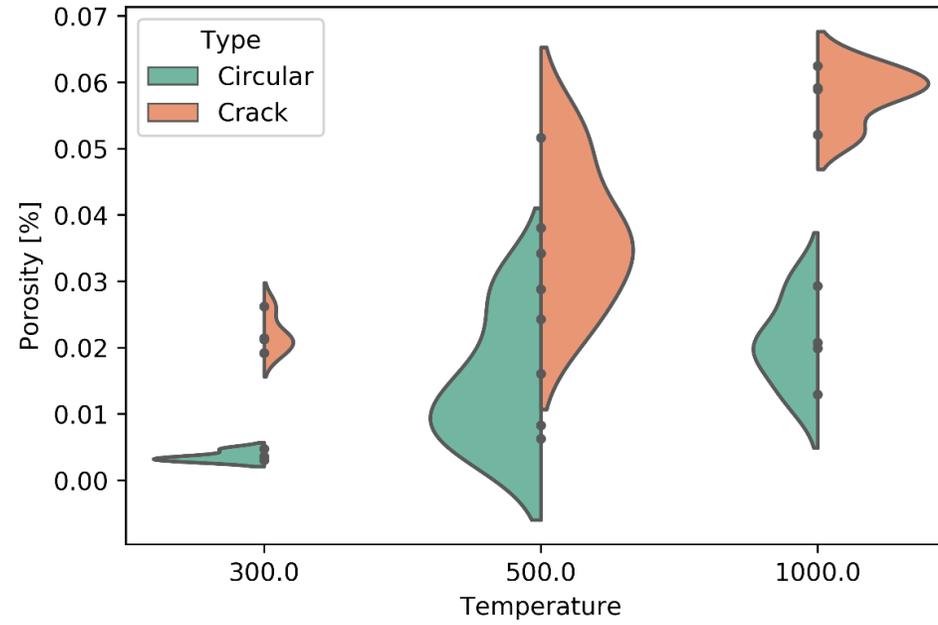
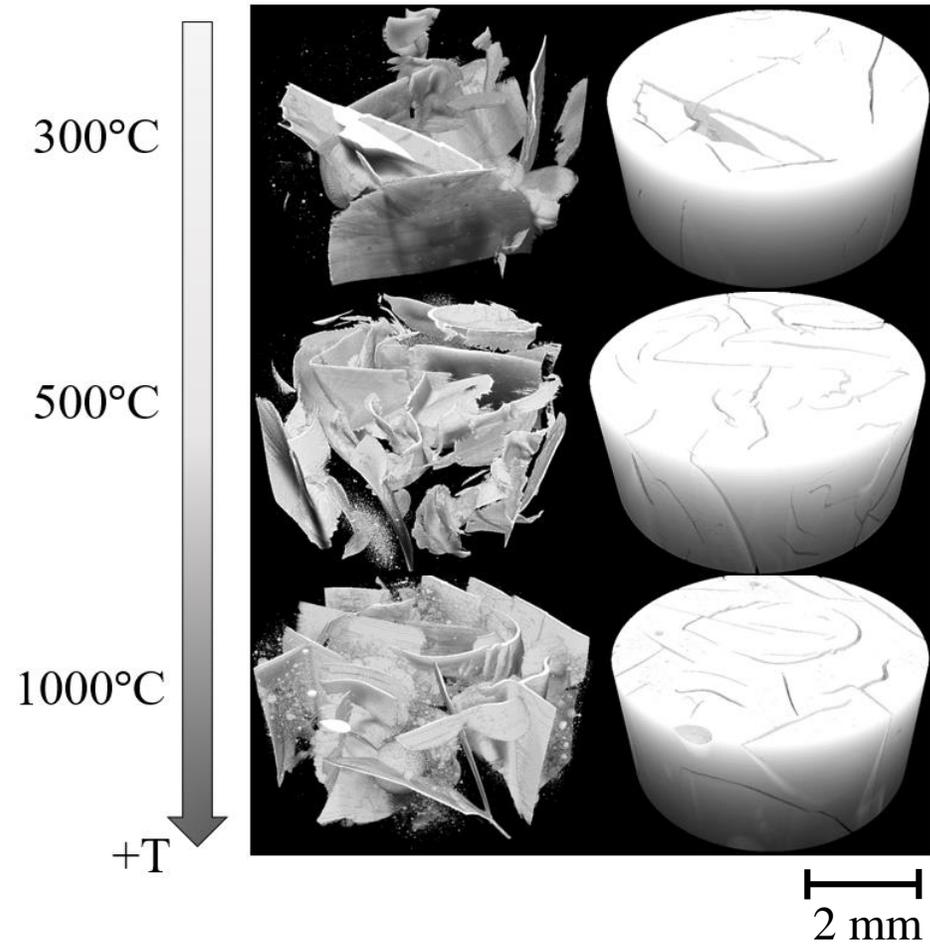


Glassy Carbon (s)
(Harris et al.)

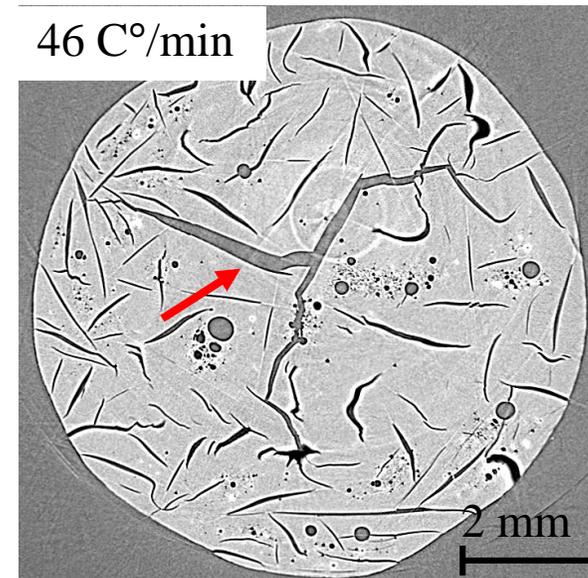
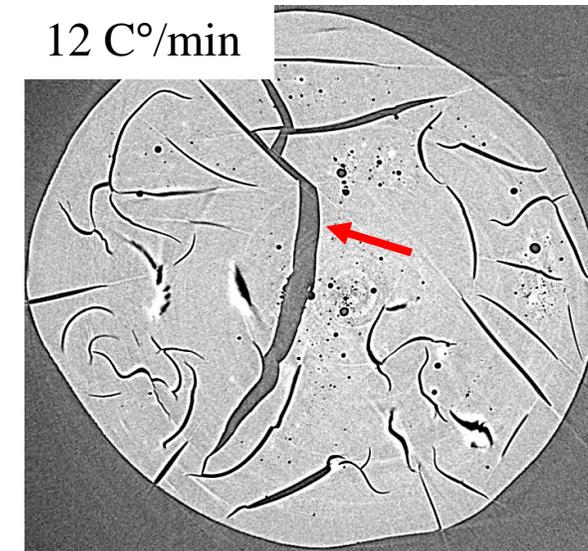
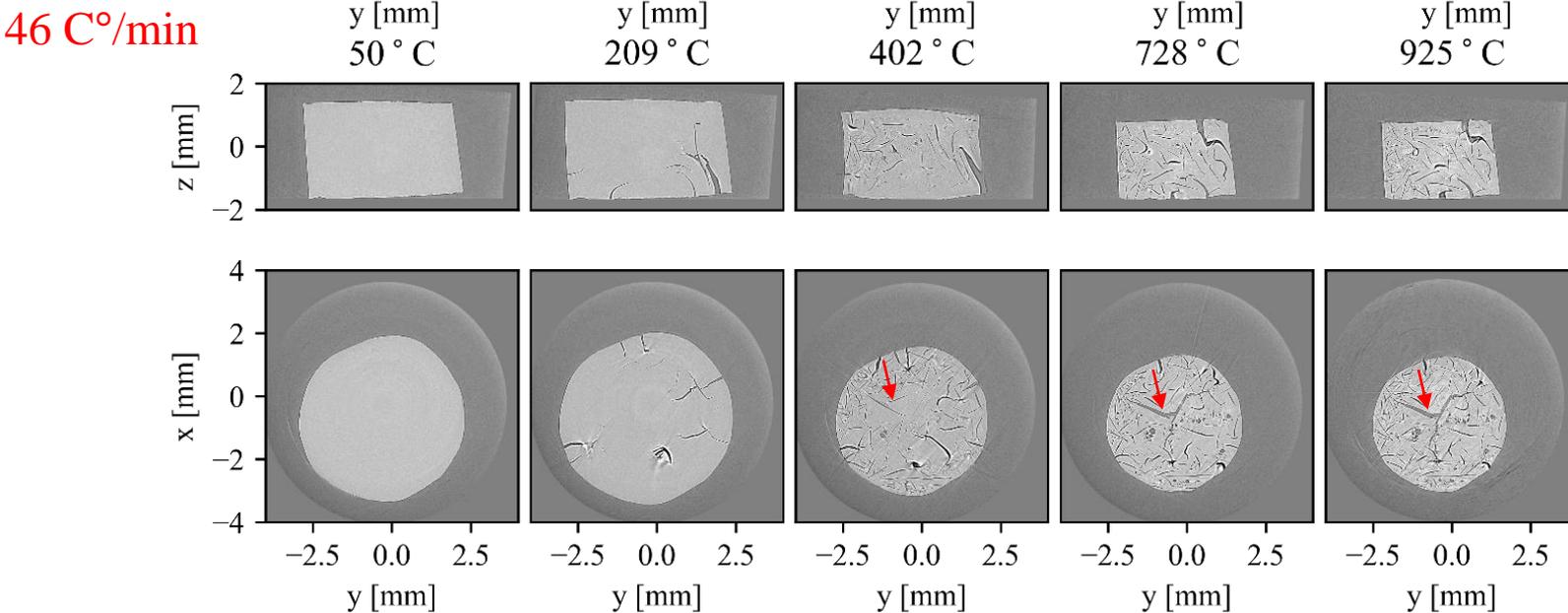
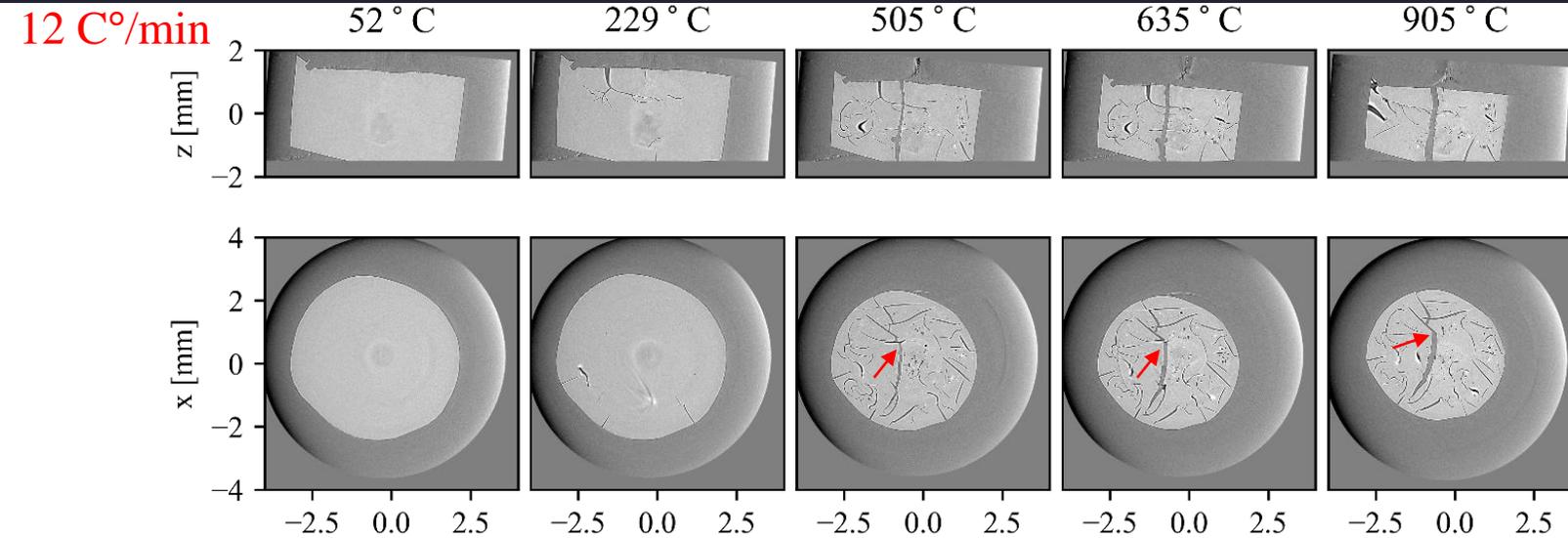


Pyrolysis products (g)

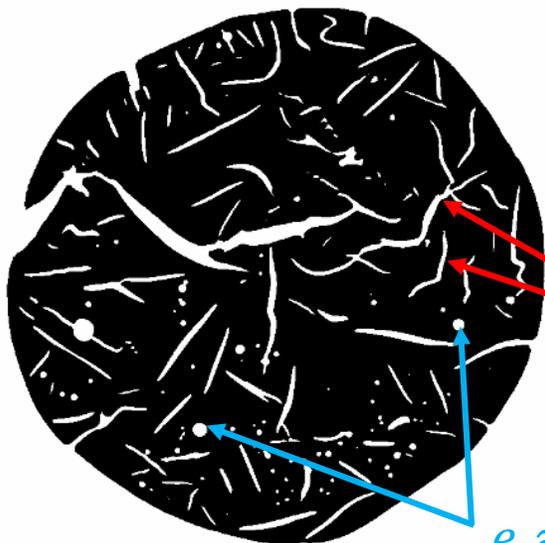
Resin ex-situ pyrolysis



Resin in-situ pyrolysis

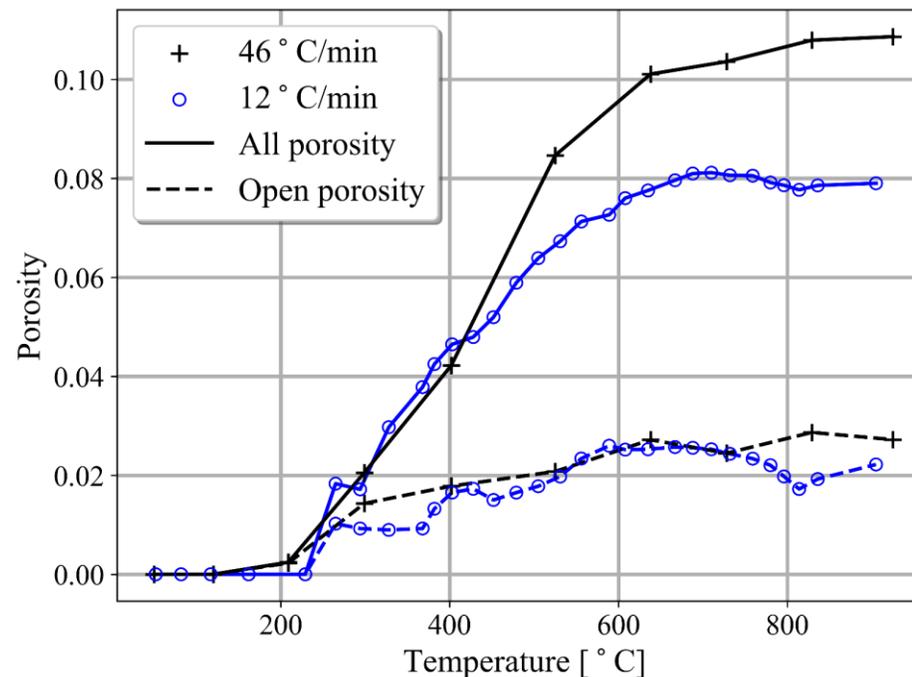
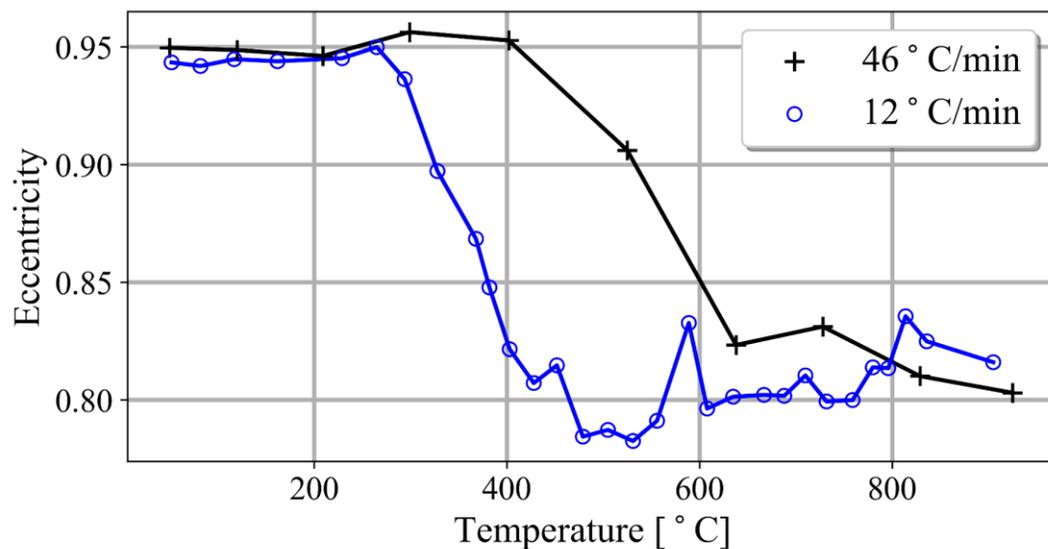
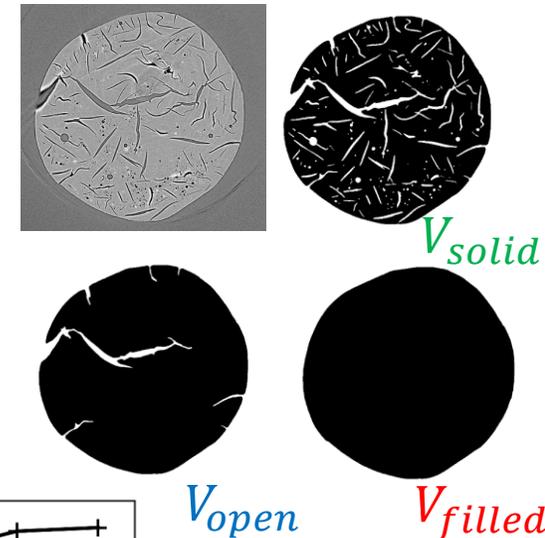


Results – Porosity

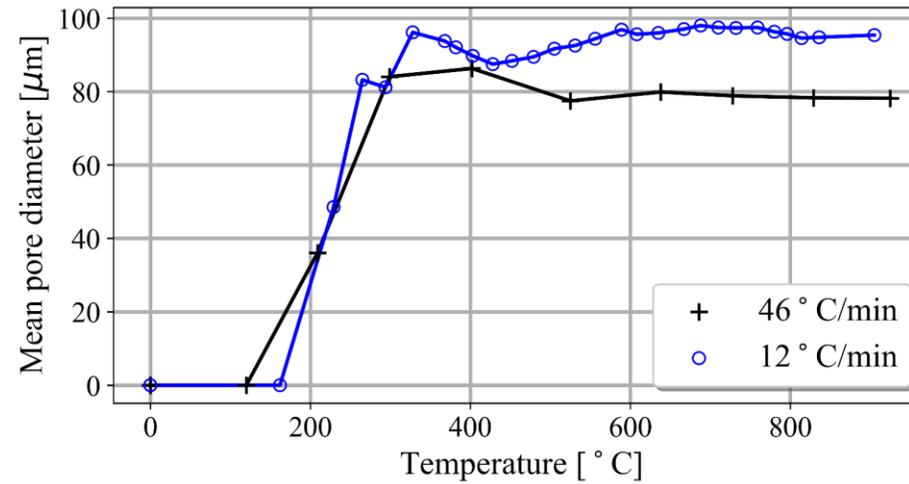
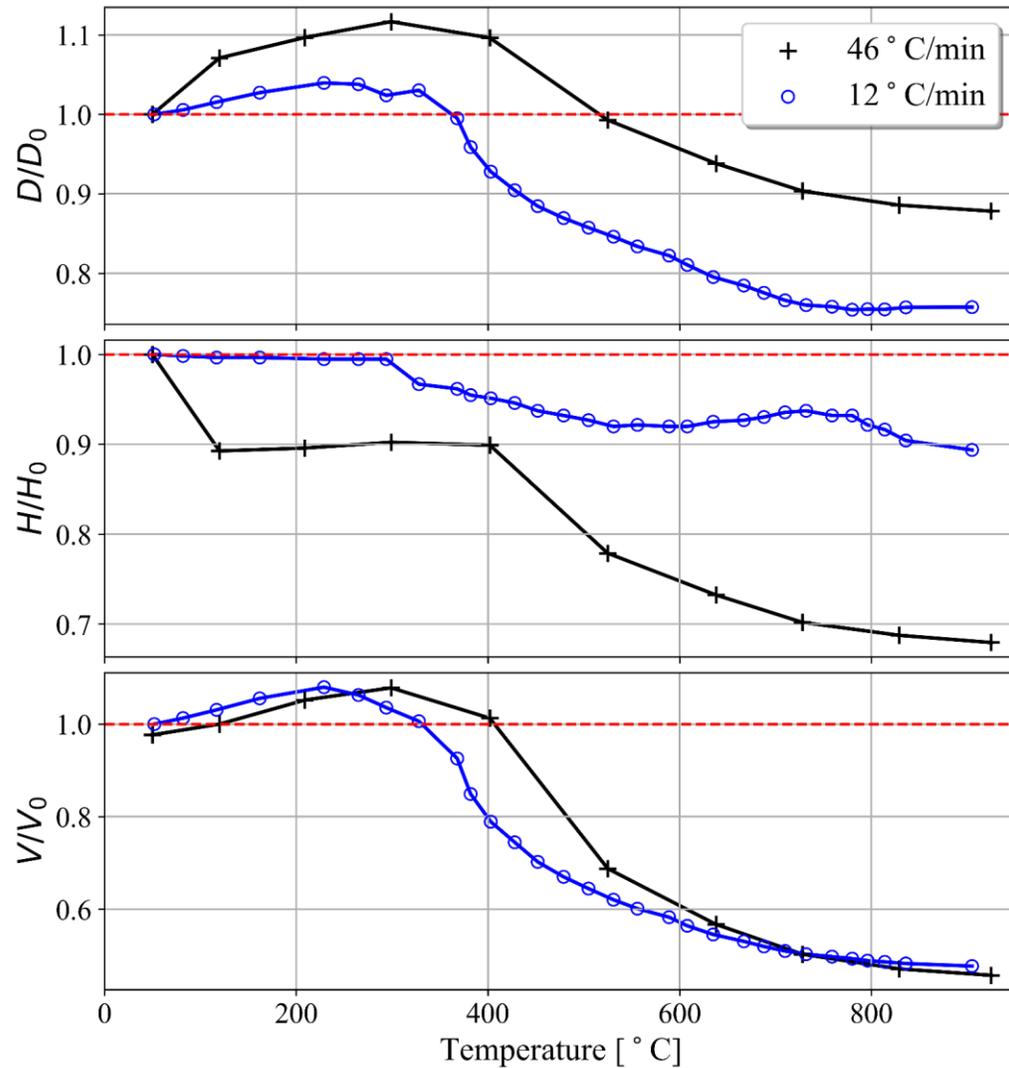


All porosity (open + closed), $\epsilon_{all} = 1 - \frac{V_{solid}}{V_{filled}}$

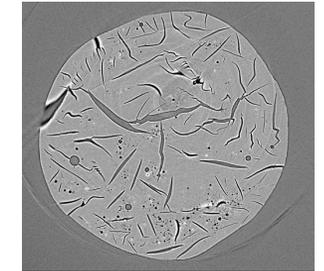
Open porosity, $\epsilon_{open} = 1 - \frac{V_{open}}{V_{filled}}$



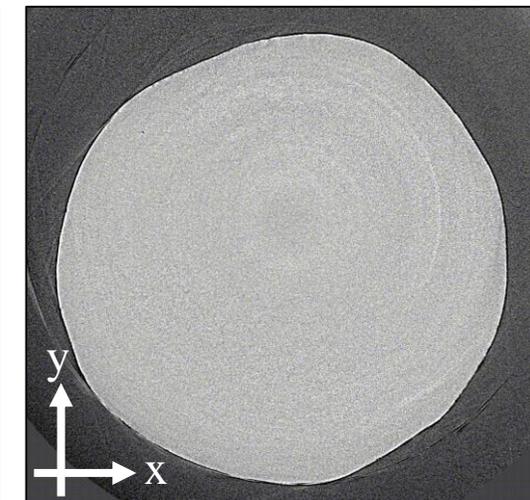
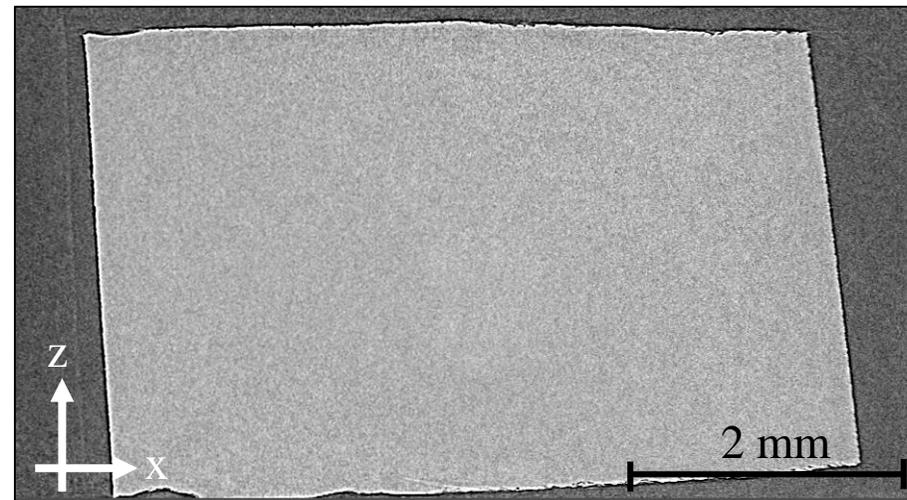
Results – Shrink and swell



Microstructure evolution of SC-1008



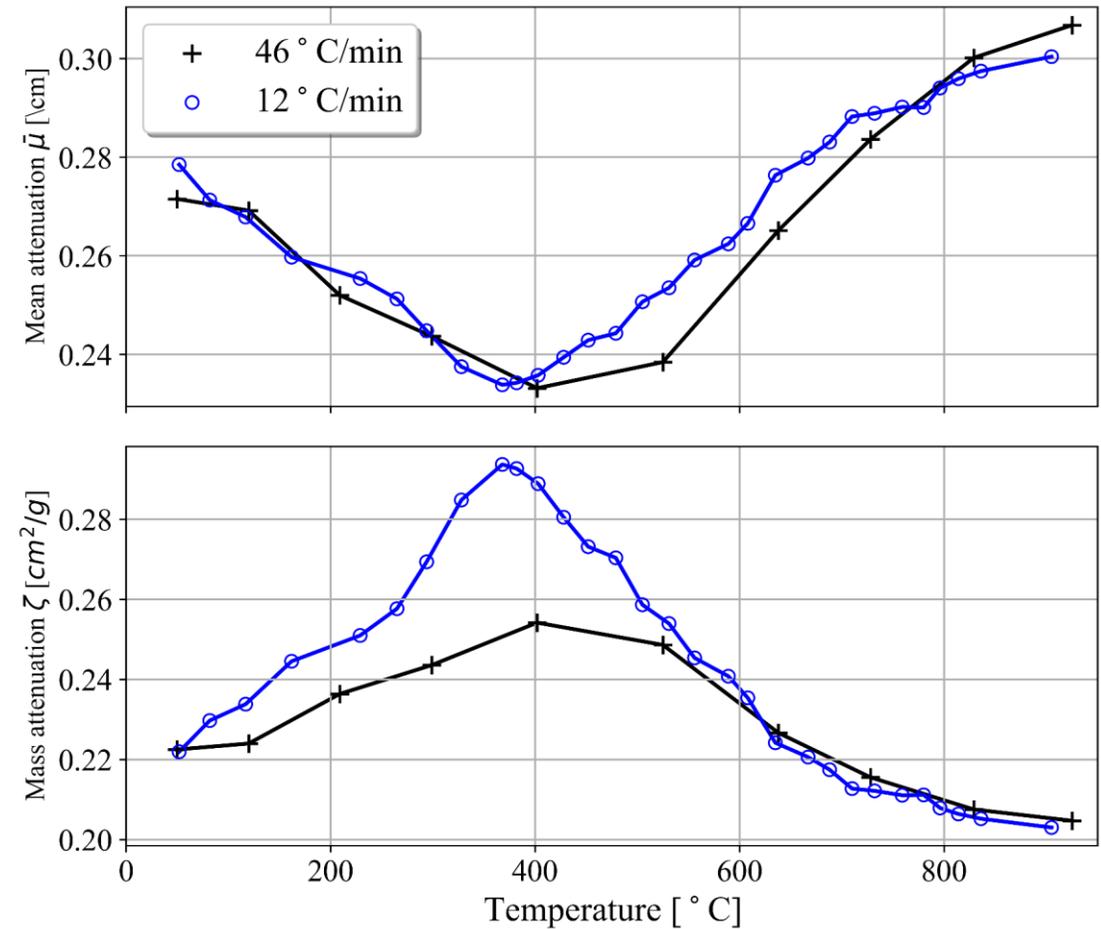
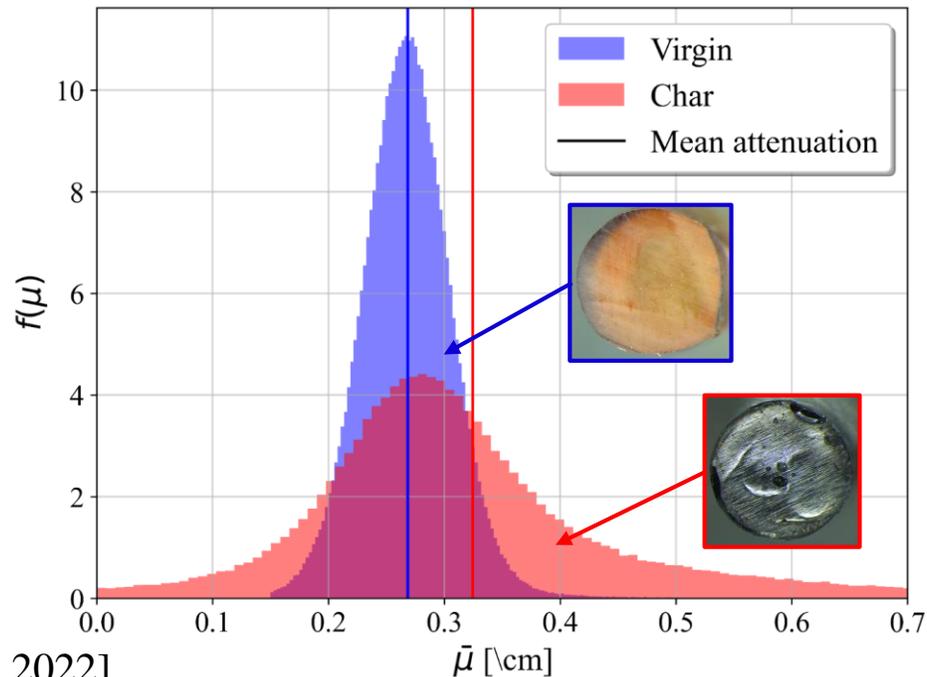
V_{solid}



Results – Density and mass loss



- (1) Mean attenuation (Beer-Lambert law), $\frac{I}{I_0} = e^{(-\mu t)}$
- (2) X-ray linear attenuation, $\mu(x, t) = \rho * \zeta$
- (3) Volume fraction of char material, $\bar{X}_c = \frac{\mu_0 - \bar{\mu}}{\mu_0 - \mu_c}$
- (4) Mean mass attenuation coefficient, $\bar{\zeta} = \frac{\bar{\mu}}{\bar{\rho}} = \frac{\mu_0 - \bar{X}_c(\mu_0 - \mu_c)}{\rho_0 - \bar{X}_c(\rho_0 - \rho_c)}$



Results – Density and mass loss

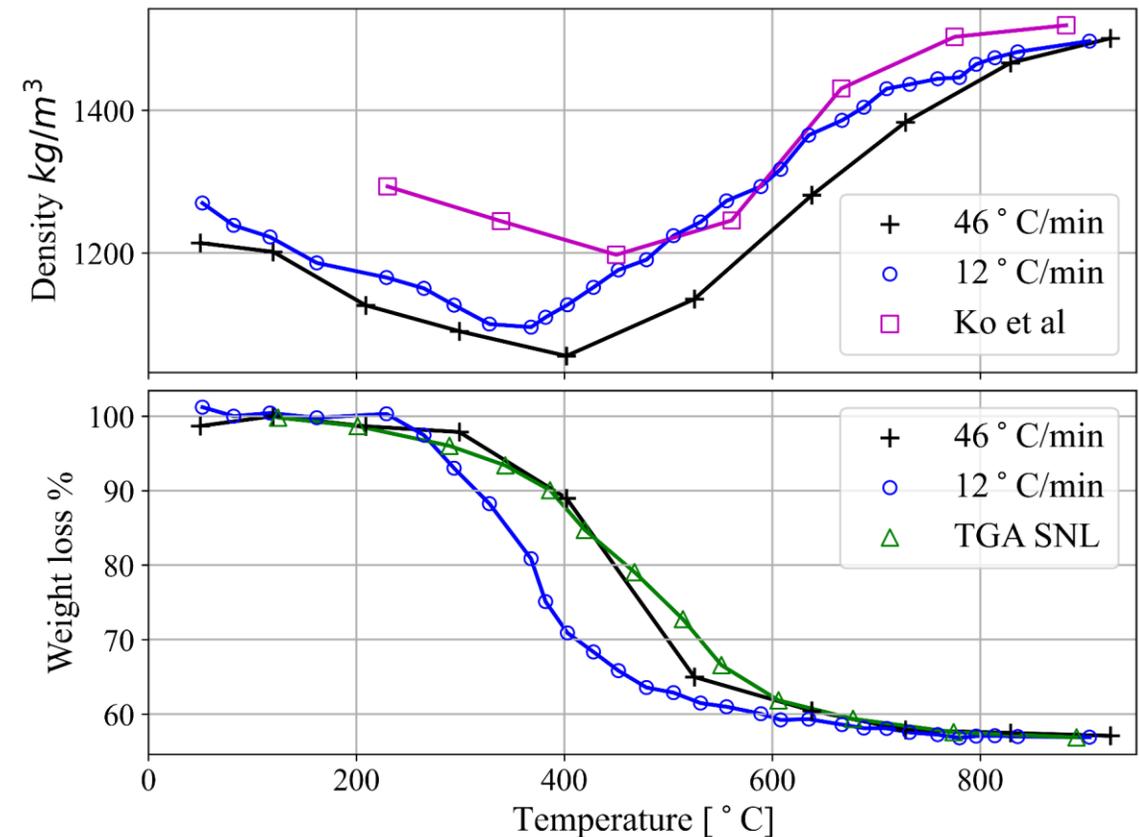


Heating rate	12 °C/min	46 °C/min
Number of tomographies	31	11
Virgin density ρ_0 [kg/m^3]	1238.84	1201.76
Virgin X-ray mass attenuation ζ_0 [cm^2/g]	0.218	0.224
Virgin mean attenuation μ_0 [$/cm$]	0.271	0.269
Char density ρ_c [kg/m^3]	1514.11	1508.66
Char X-ray mass attenuation ζ_c [cm^2/g]	0.201	0.204
Char mean attenuation μ_c [$/cm$]	0.304	0.308
Volatile matter $Y_p = 1 - m_c/m_0$ [wt%]	43.2	43.2
Extent of volume shrinkage $\gamma = V_c/V_0$ [%]	46.5	45.2
Temperature at 40% mass loss [°C]	595	630

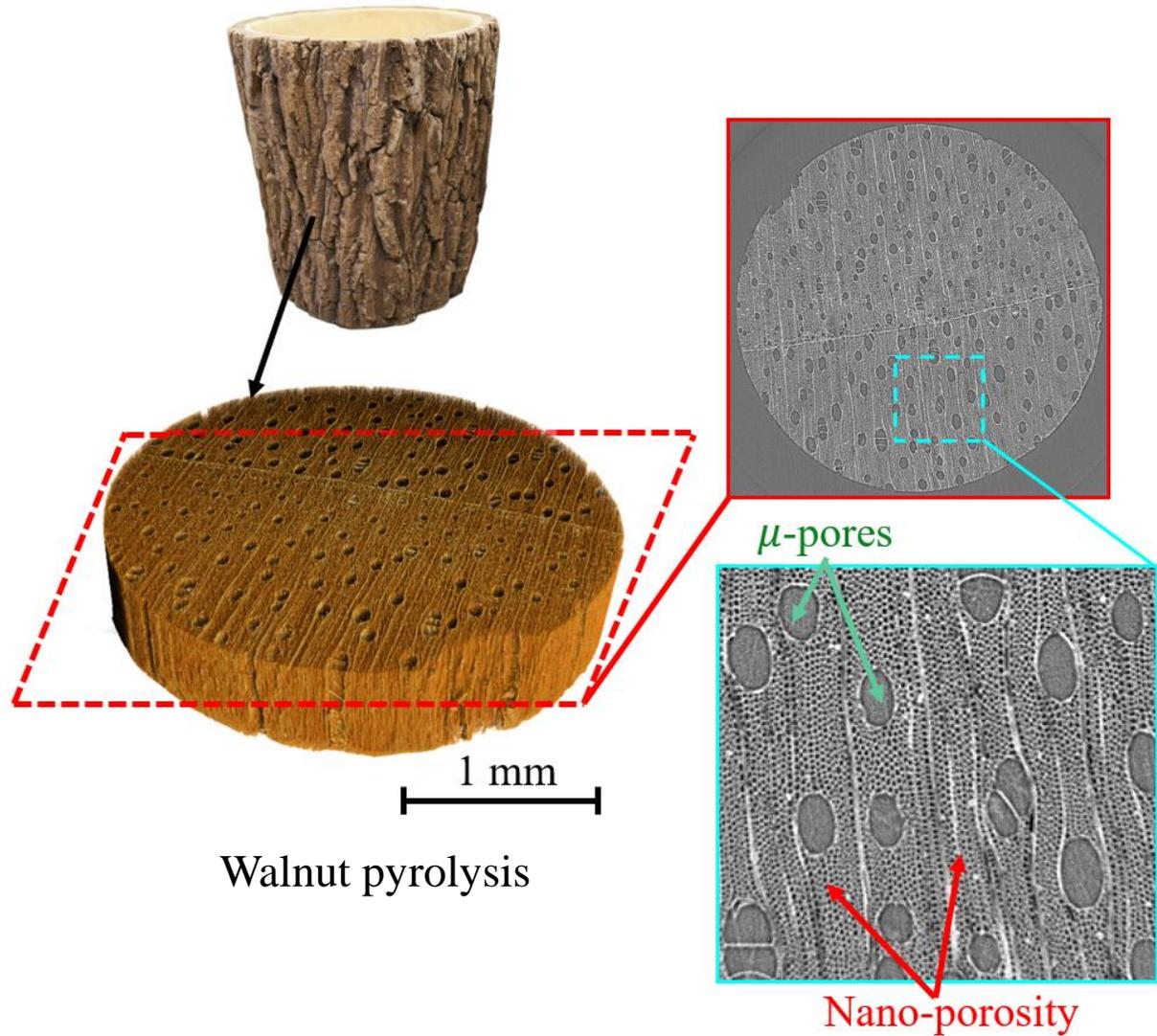
Morphological evolution SC-1008



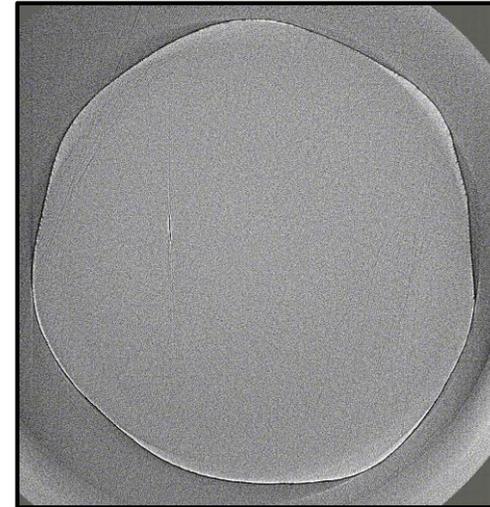
3 mm



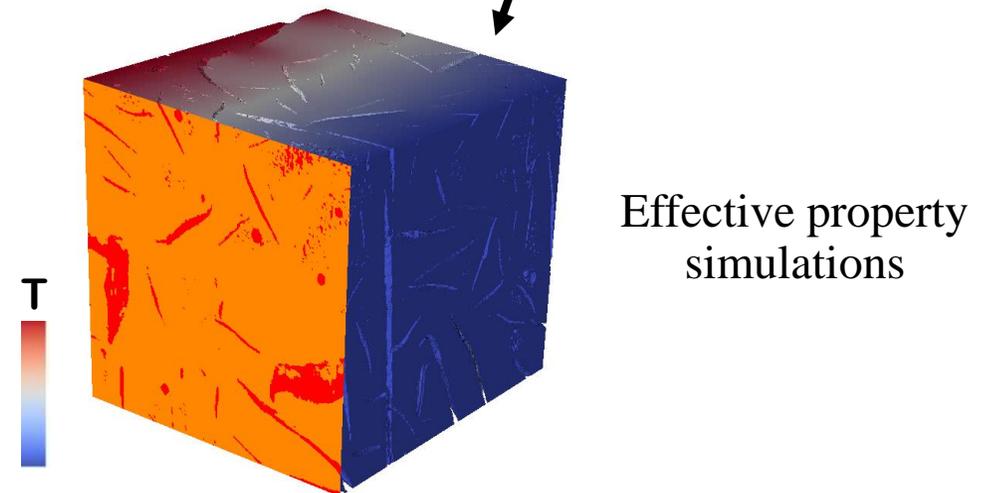
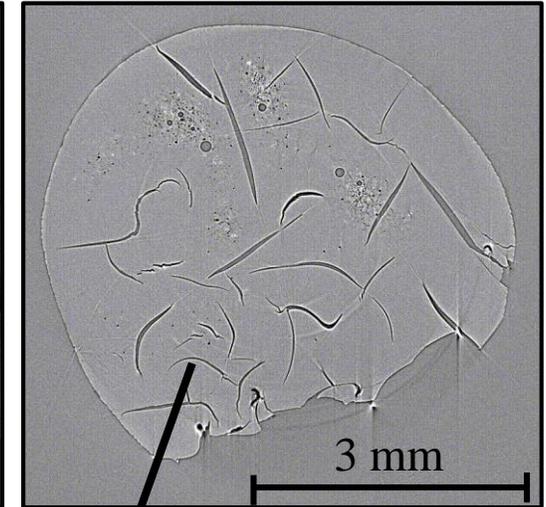
Outlook



Oxidation of virgin



Oxidation of pre-char



Acknowledgements



Sandia National Laboratories is a multimission laboratory managed and operated by National Technology & Engineering Solutions of Sandia, LLC, a wholly owned subsidiary of Honeywell International Inc., for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-NA0003525.



Supplementary



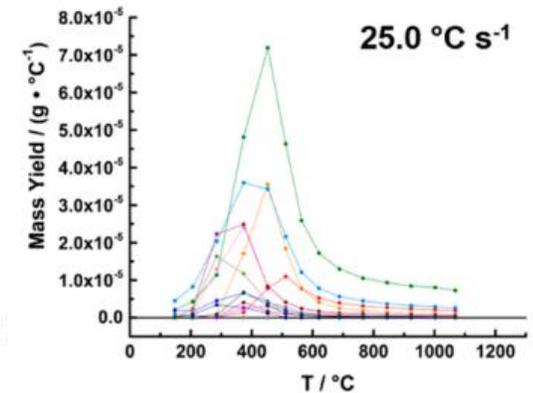
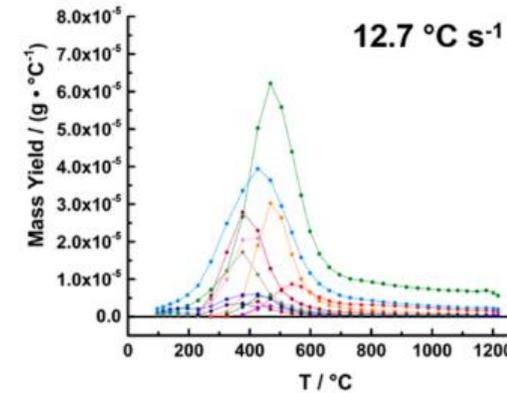
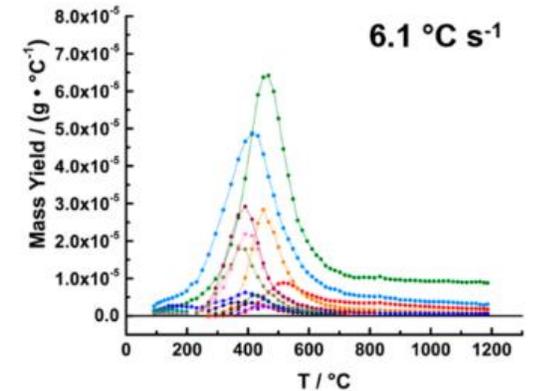
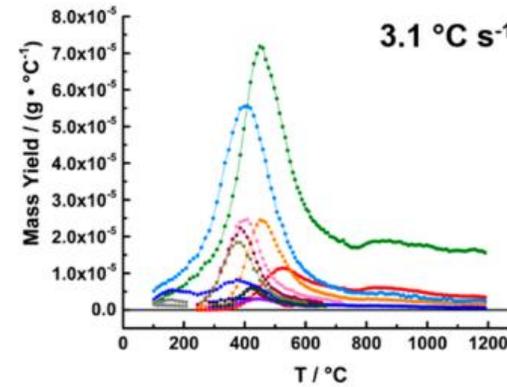
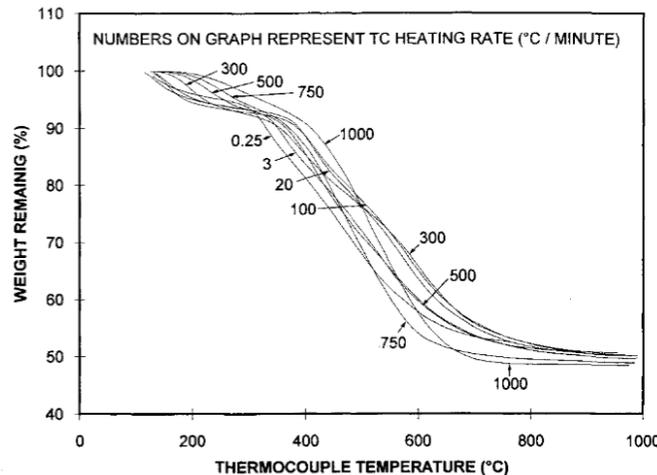
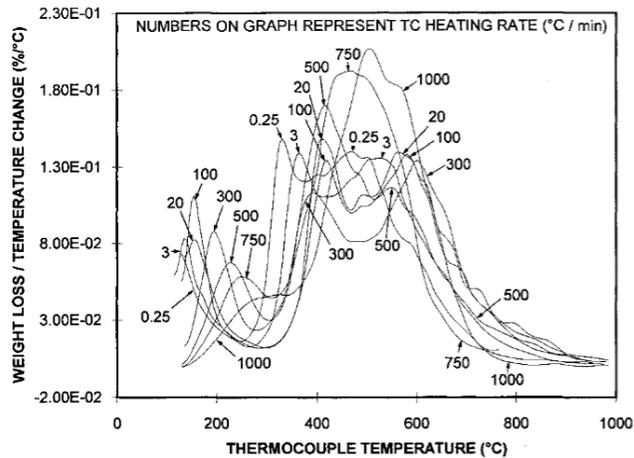
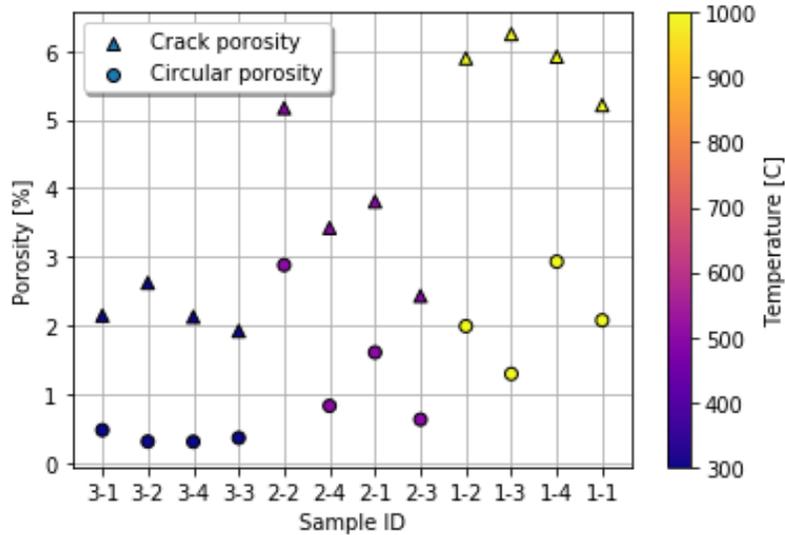
$$\mu(x, t) = \rho \zeta$$

$$\int \mu dV = \int \zeta \rho dV$$

$$\int \mu dV = \bar{\zeta} * \int \rho dV$$

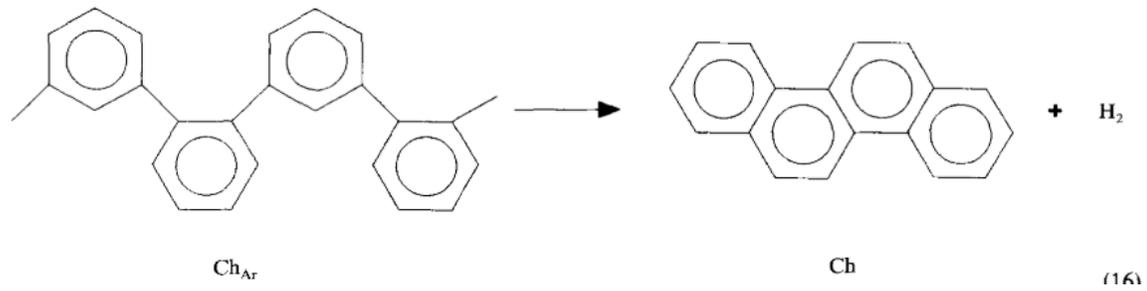
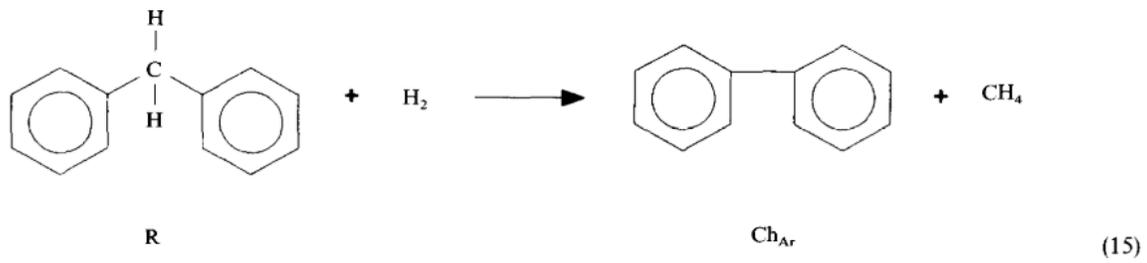
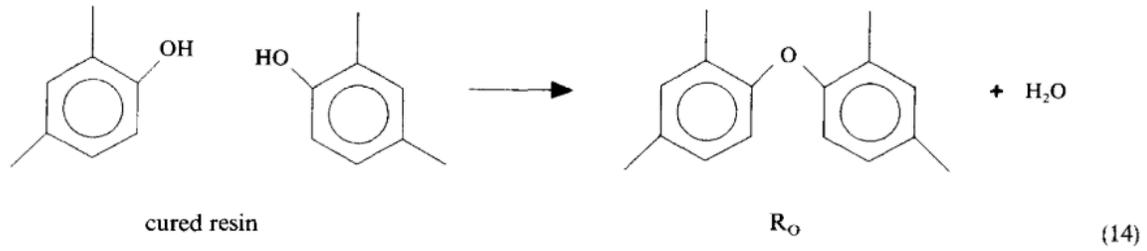
$$\bar{\mu} = \bar{\zeta} * m$$

$$\rightarrow m \approx \bar{\mu} / \bar{\zeta}$$



- H₂
- CH₄
- CO
- CO₂
- H₂O
- 1-propanol
- 2-propanol
- Phenol
- Xylene
- Cresol
- Toluene
- Benzene
- Trimethyl Phenol
- Dimethyl Phenol

Supplementary



Supplementary



Approximating temperature gradient:

$$\rho = 1238.84 \frac{kg}{m^3}$$

$$C_p = 1180 \frac{J}{kg * K}$$

$$V = 5.89E - 7$$

$$\frac{dT}{dt} = 4.75K/s$$

$$A_s = 8.64E - 5$$

$$q'' = \rho C_p V * \frac{dT}{dt} * 1/A_s = 4734.34 W/m^2$$

$$T^* = \frac{k(T - T_i)}{q''L} = 0.5$$

$$\Delta T = 0.5 * \frac{q''L}{k}$$

$$\Delta T = 0.5 * \frac{\left(\frac{0.473W}{cm^2}\right) * 0.25cm}{\frac{0.001W}{cmK}} = \mathbf{60K}$$

Largest temperature gradient being 60K

Approximating time for quasi-steady state:

$$Fo = 0.5$$

$$t = 0.5 * \frac{L^2}{\alpha}$$

$$t = 0.5 * \frac{(2.5mm)^2}{\frac{0.167mm^2}{s}} = \mathbf{18 sec}$$

Quasi steady state reached quickly compared to timescale of each tomography (~1 tomo per 1.5 min)

Supplementary



Virgin

300C

500C

1000C

